

Achieving Extremely Low Effluent TN and TP Concentration at Very Low SRTs Using IFAS Technology

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Purposes

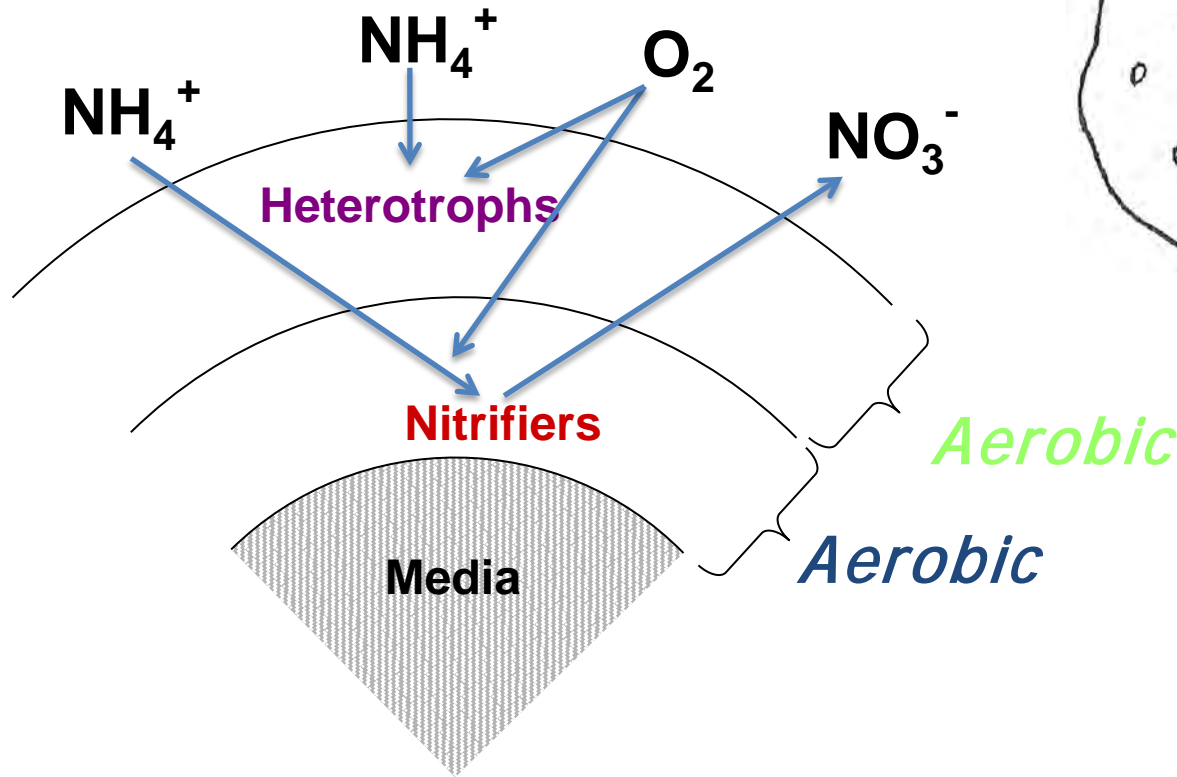
- Investigate the ability of IFAS processes with extremely short SRTs to remove total nitrogen
- Investigate ability of development of PAOs in the sludge blanket in the secondary clarifiers
- Summarize the evolving understanding of the Bio-P removal mechanism in the sludge blanket

Design and Operation of an IFAS

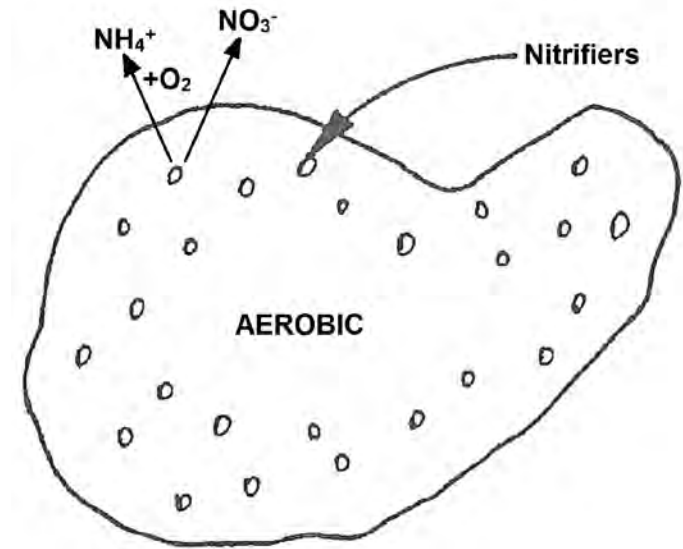
IFAS design is easy !

1. Very similar to Conventional Activated Sludge System
2. However, nitrification exists in both suspended and attached biomass
 - High load (low SRT) – nitrification in biofilm dominates
 - Low load (high SRT) – nitrification in suspension plays a greater role
 - Higher nitrification rates at a given SRT in CAS – seeding effect from attached biomass
 - Higher DO Setpoint (3-4 mg/L)

Nitrification in IFAS



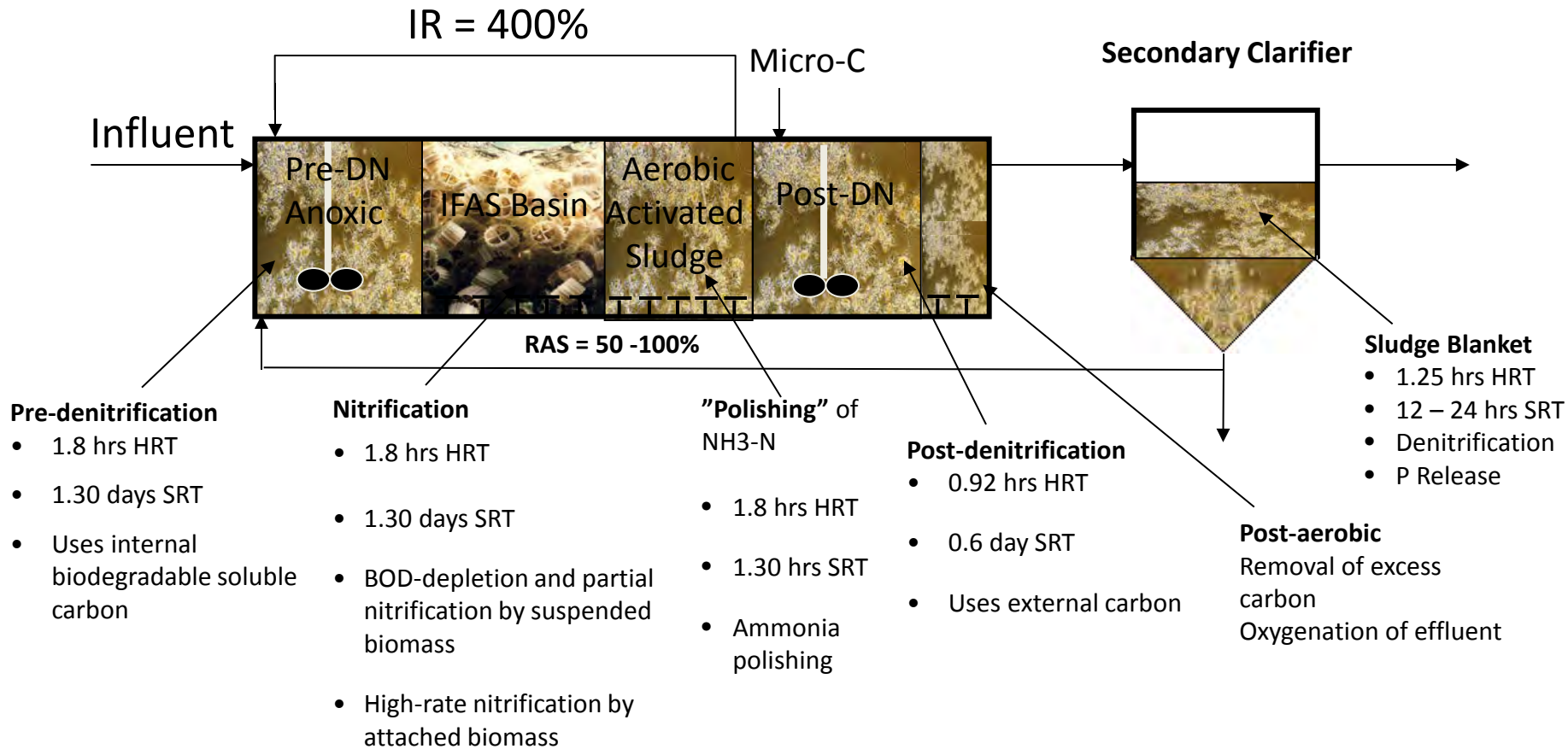
Nitrification on carrier



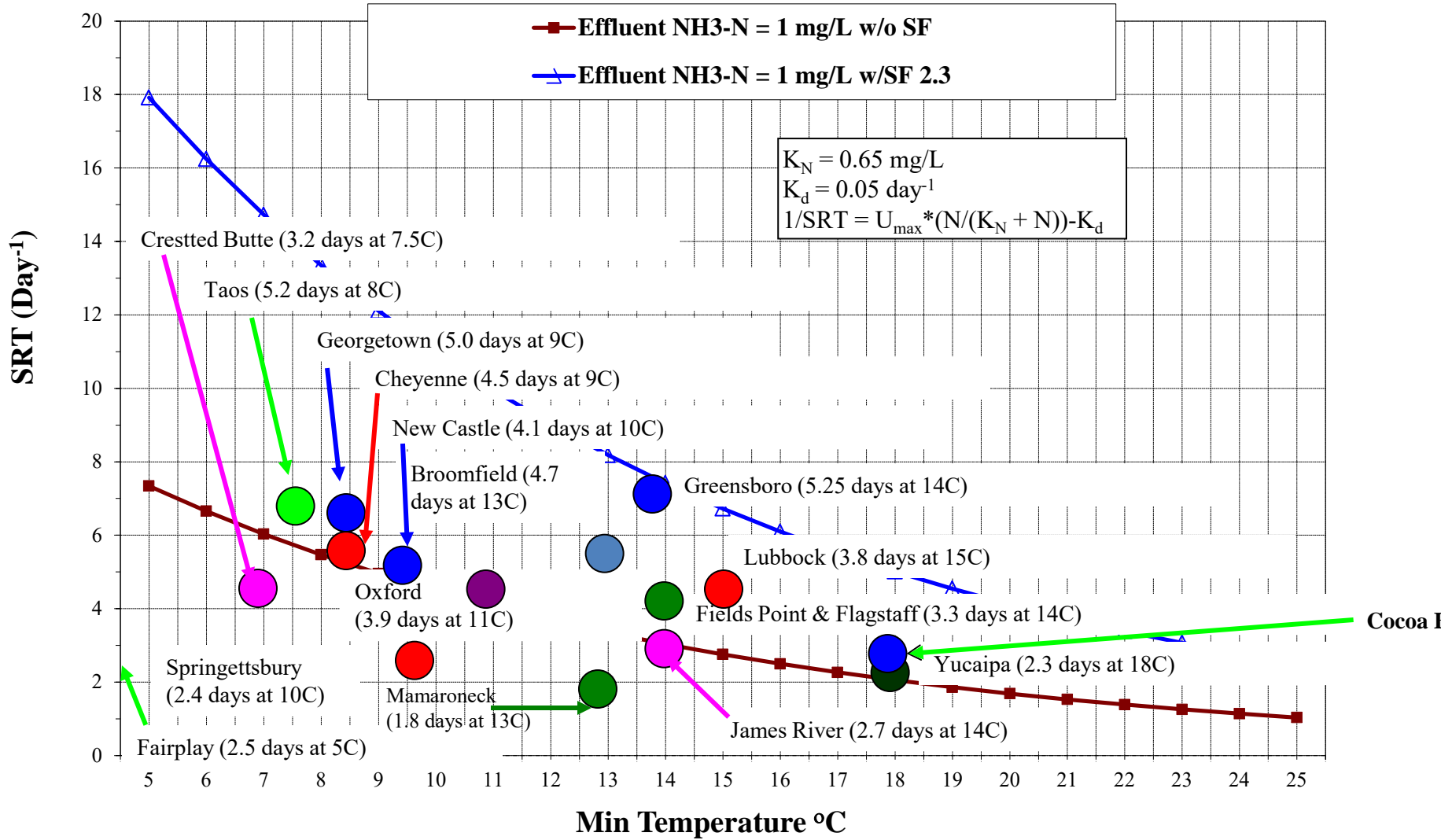
Nitrification on Suspended Biomass

Simultaneous
in biofilm

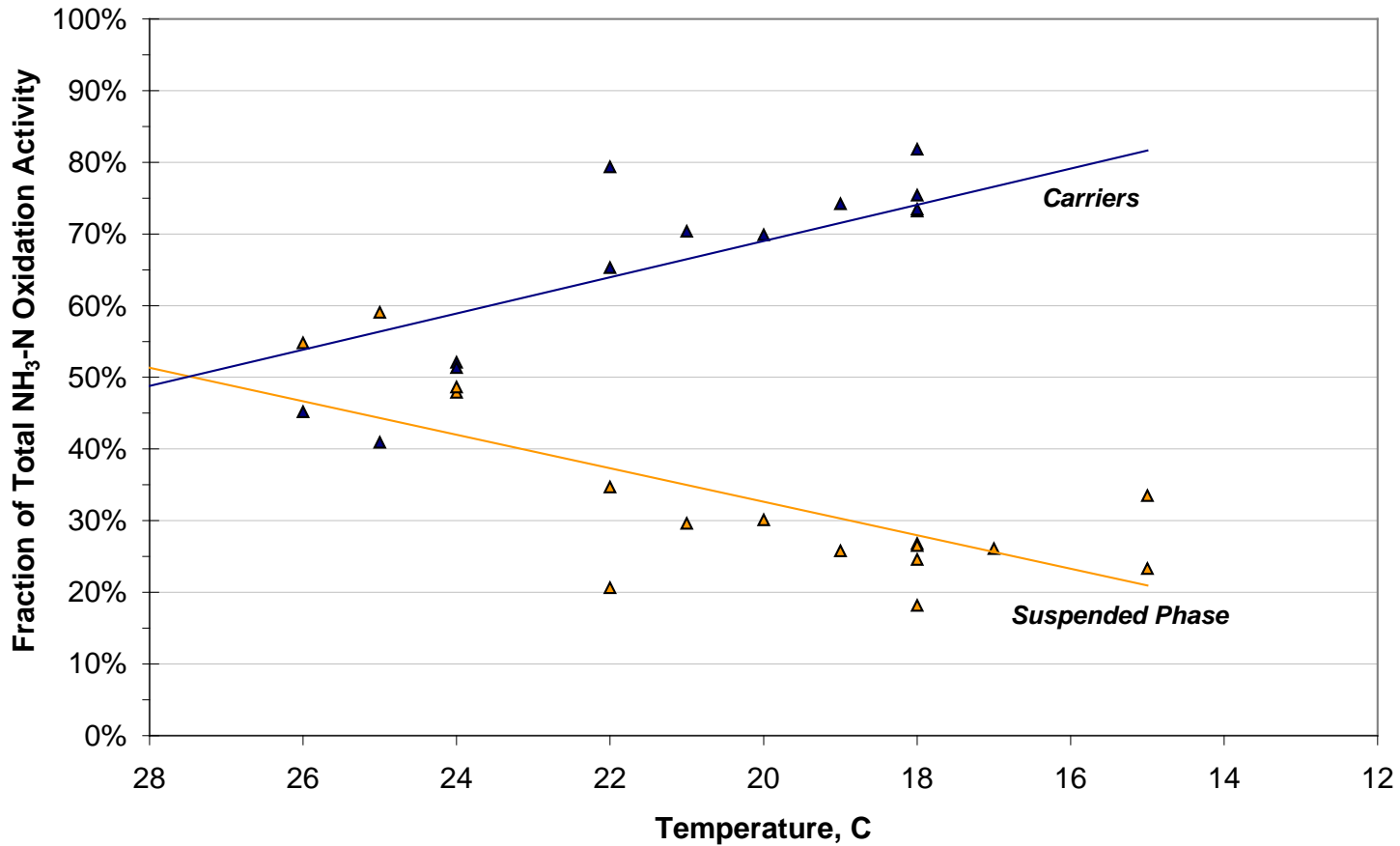
IFAS Process Configuration



SRT For Nitrification



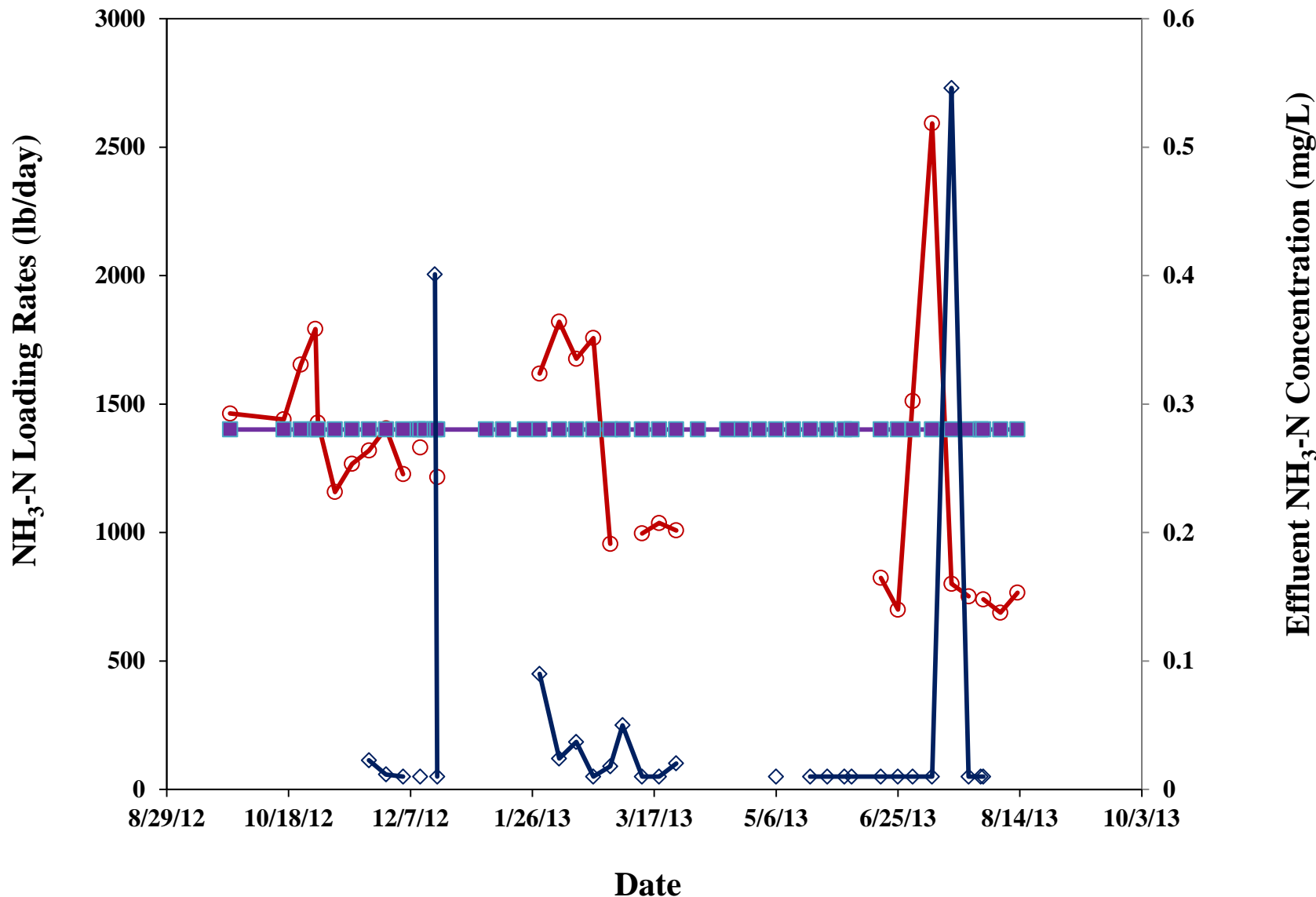
Seasonal Shift in Ammonia Oxidation Activity



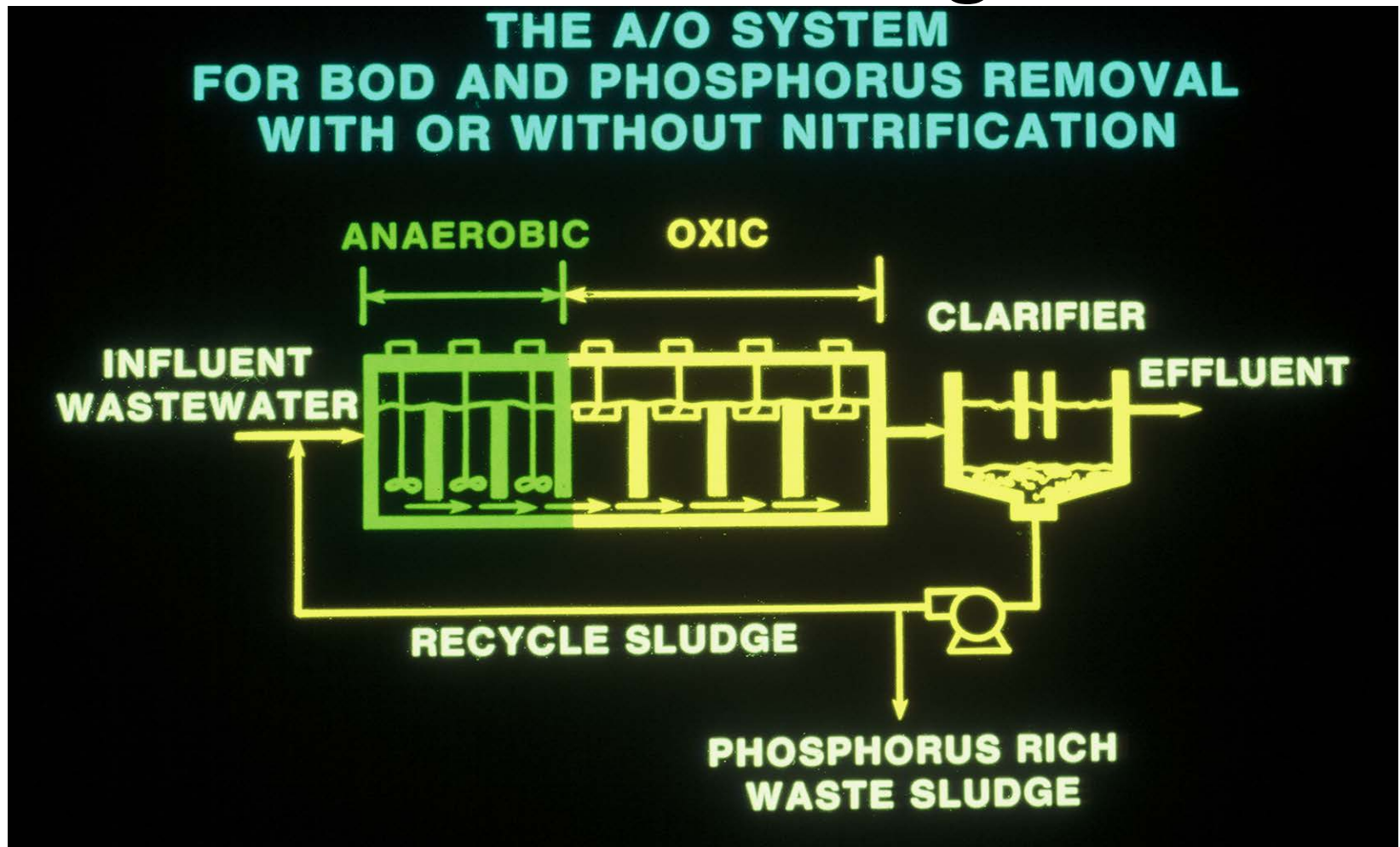
○ Influent NH₃-N Loading

■ Design NH₃-N Loading

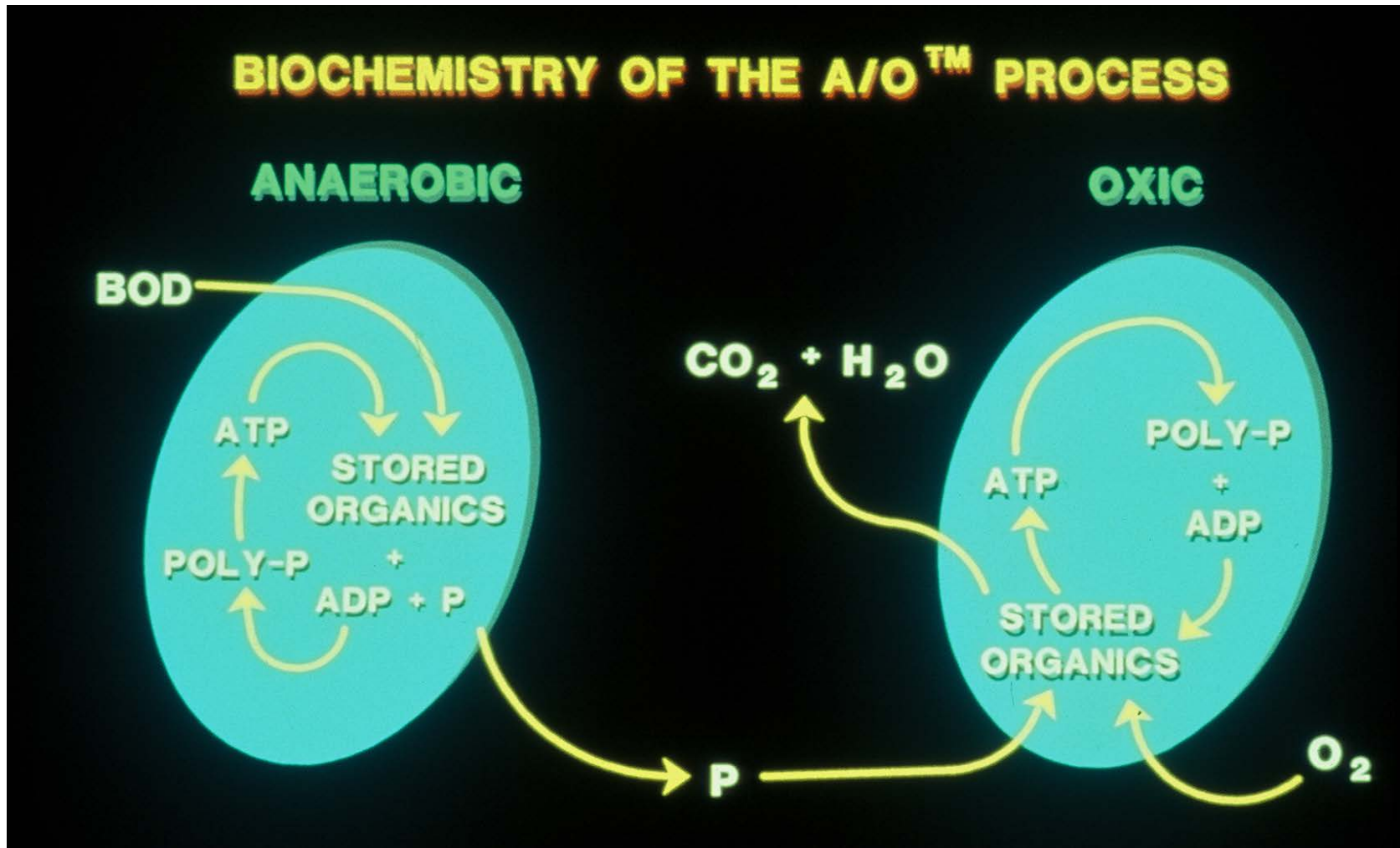
◇ Effluent NH₃-N Concentrations

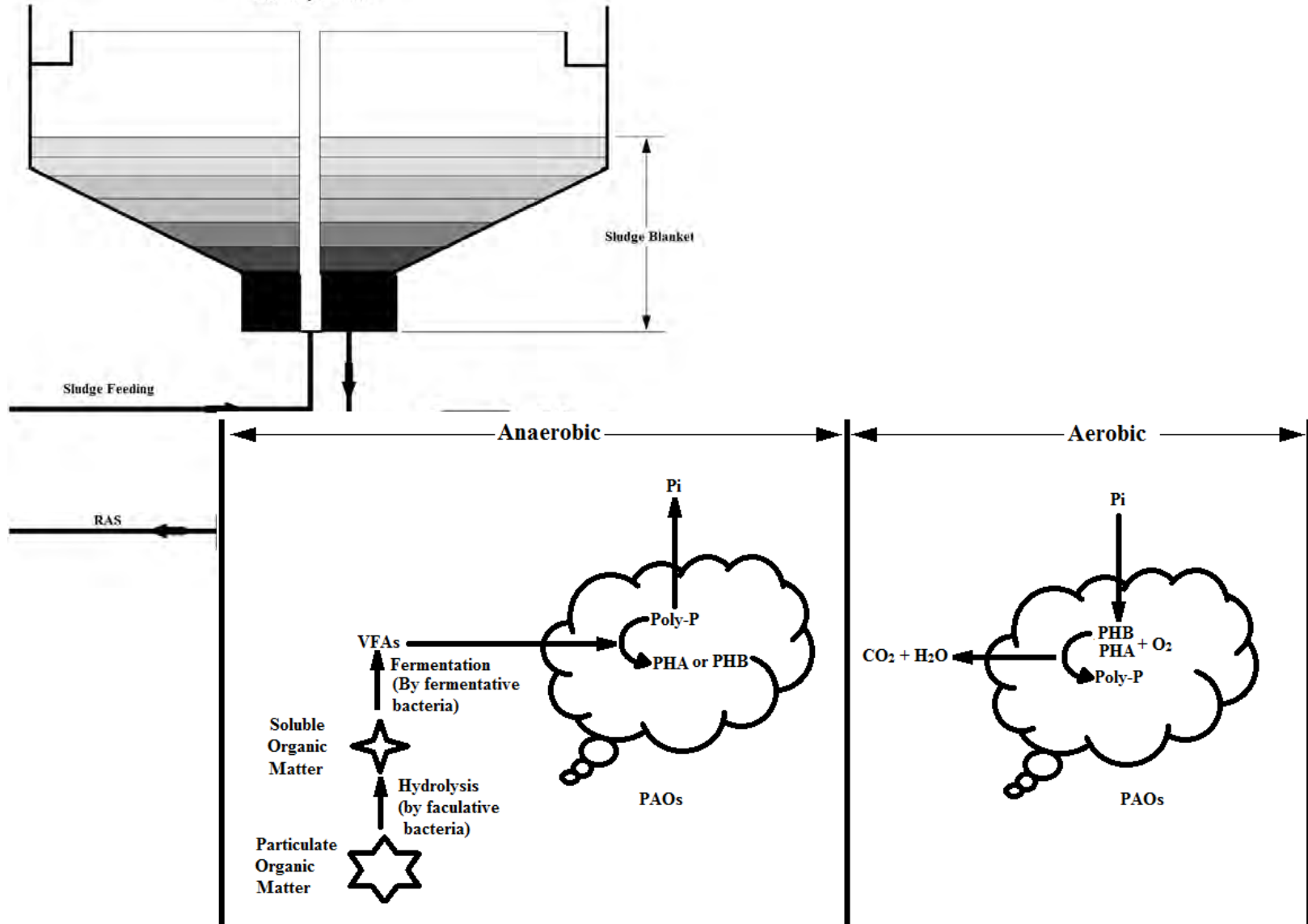


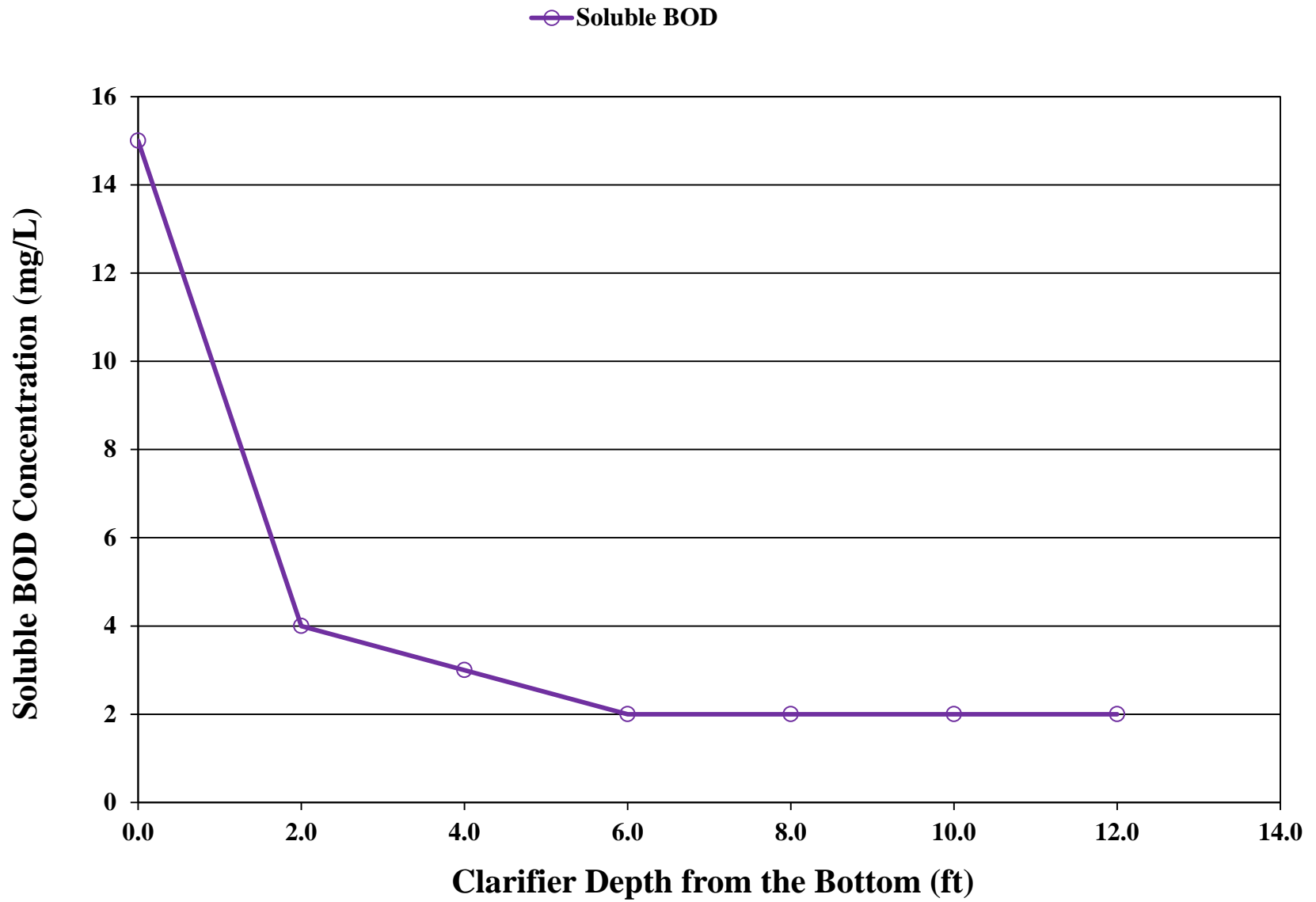
The Classic EBPR Configuration

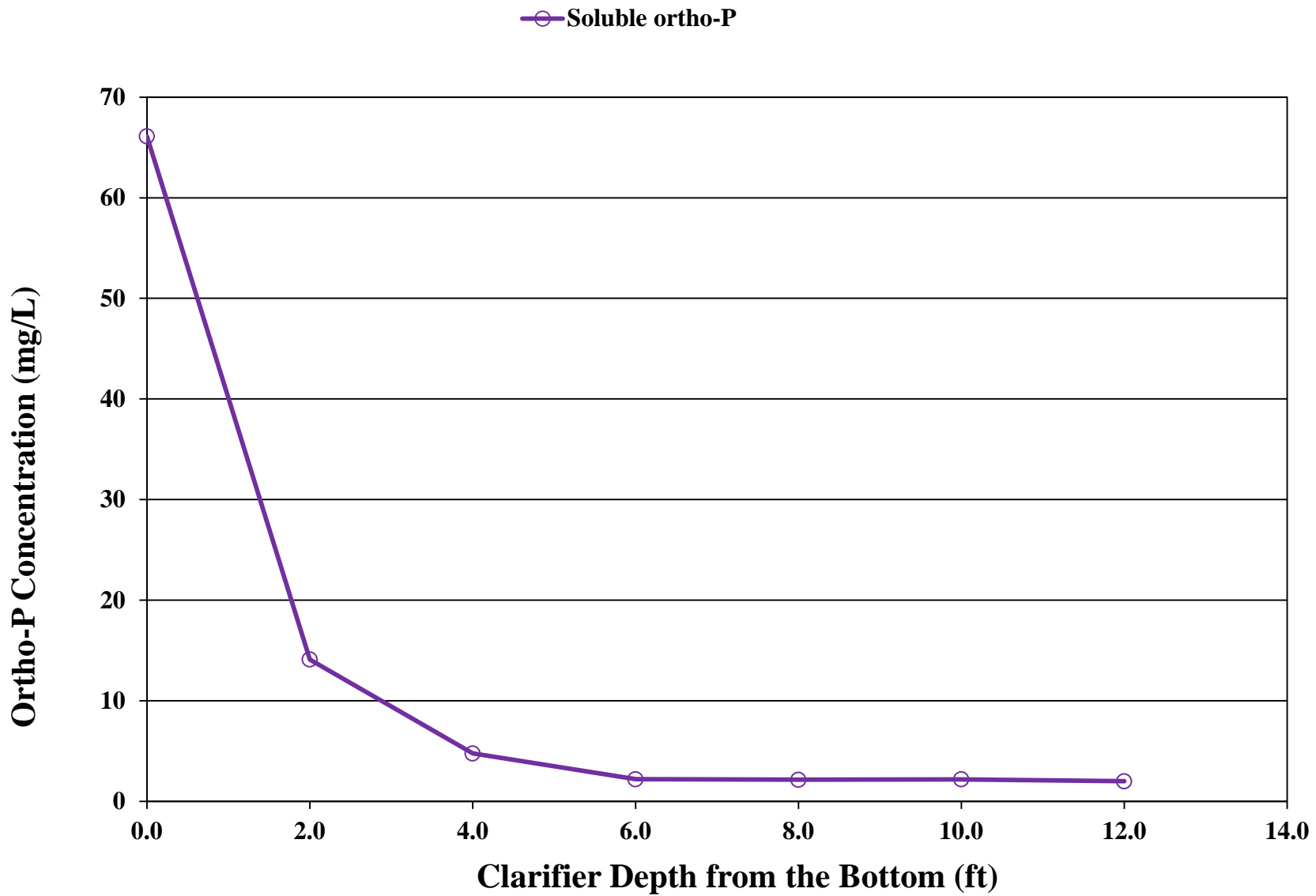


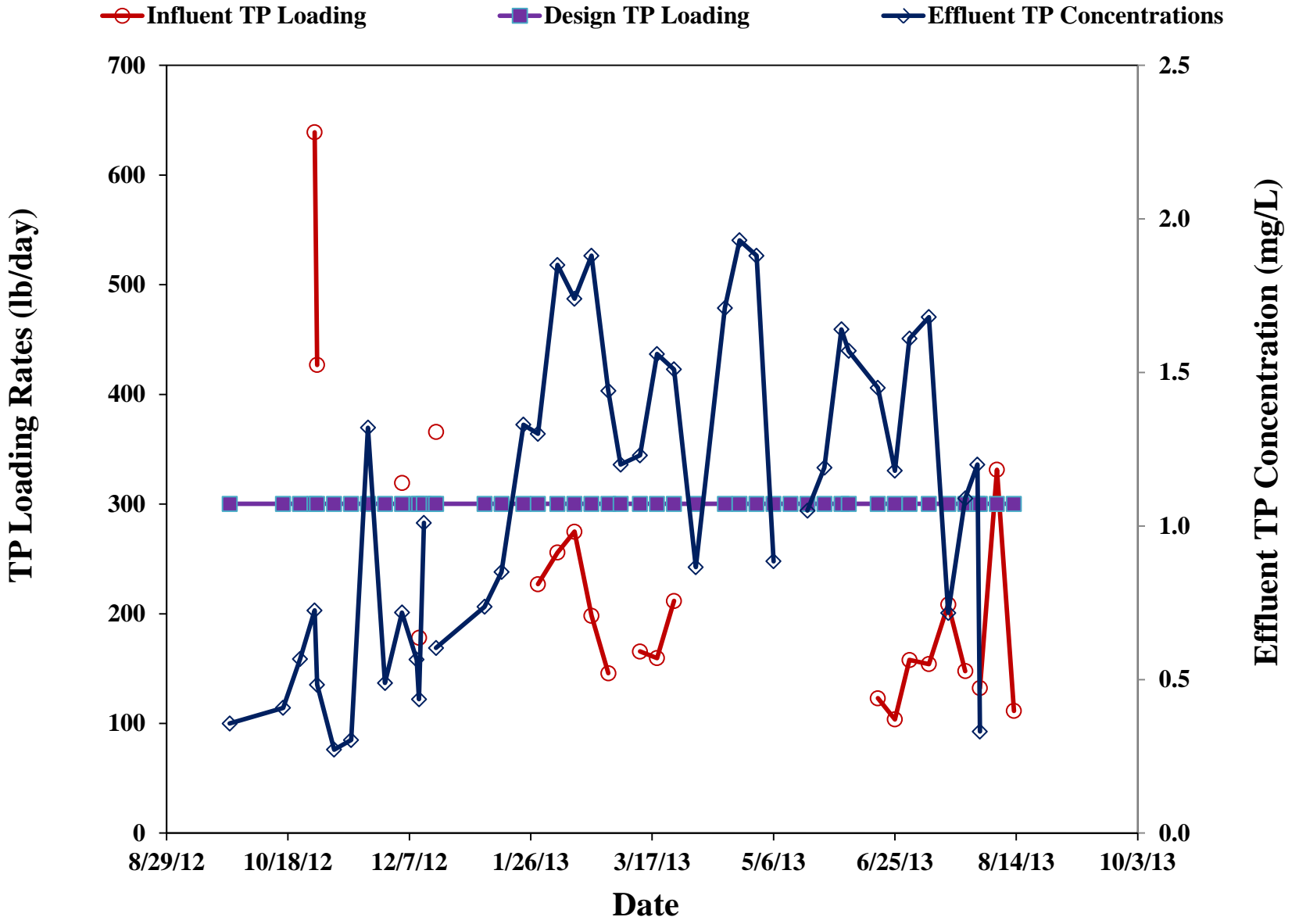
Bio-P Removal Mechanism









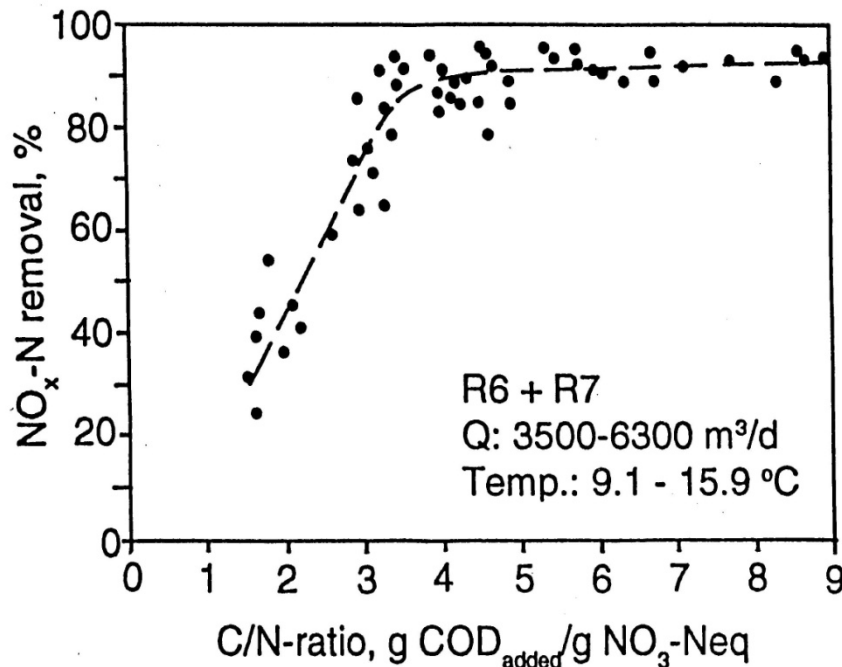


Post-Denitrification

$$r_{DN} = k * [S_{COD}/(S_{COD} + K_{S,COD})] * [S_{NO3}/(S_{NO3} + K_{S,NO3})]$$

What this equation tells us?

- Availability of carbon source is normally the limiting factor ($K_S = 2,5 - 5$ mg SBCOD/l) when C/N ratio is less 3 mg COD/mg
- If very low effluent NO_3-N (< 1-3 mg/l) is required, NO_3-N is the limiting factor and determining the DN-rate ($K_S = 0,5-2,0$ mg NO_3/l)

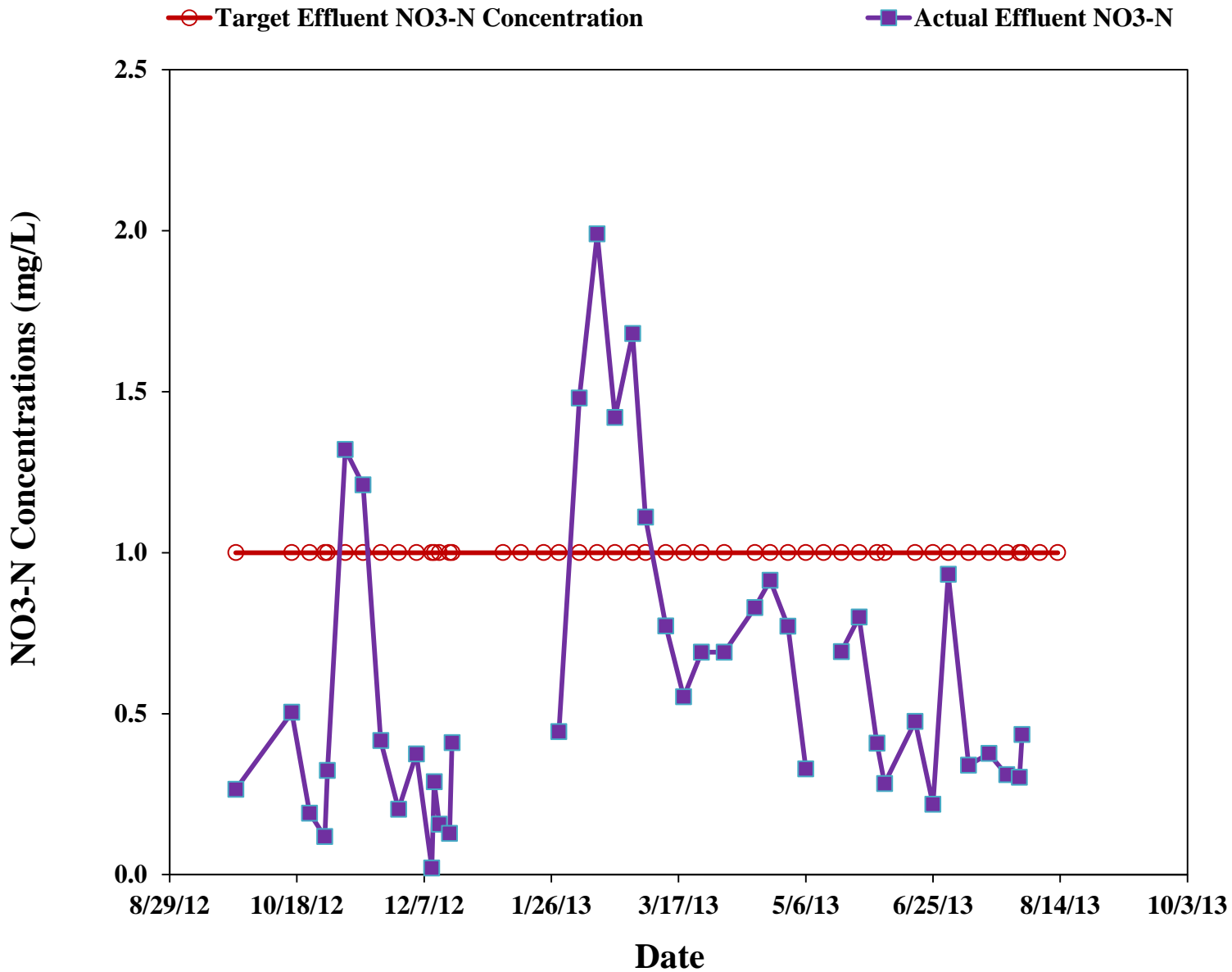


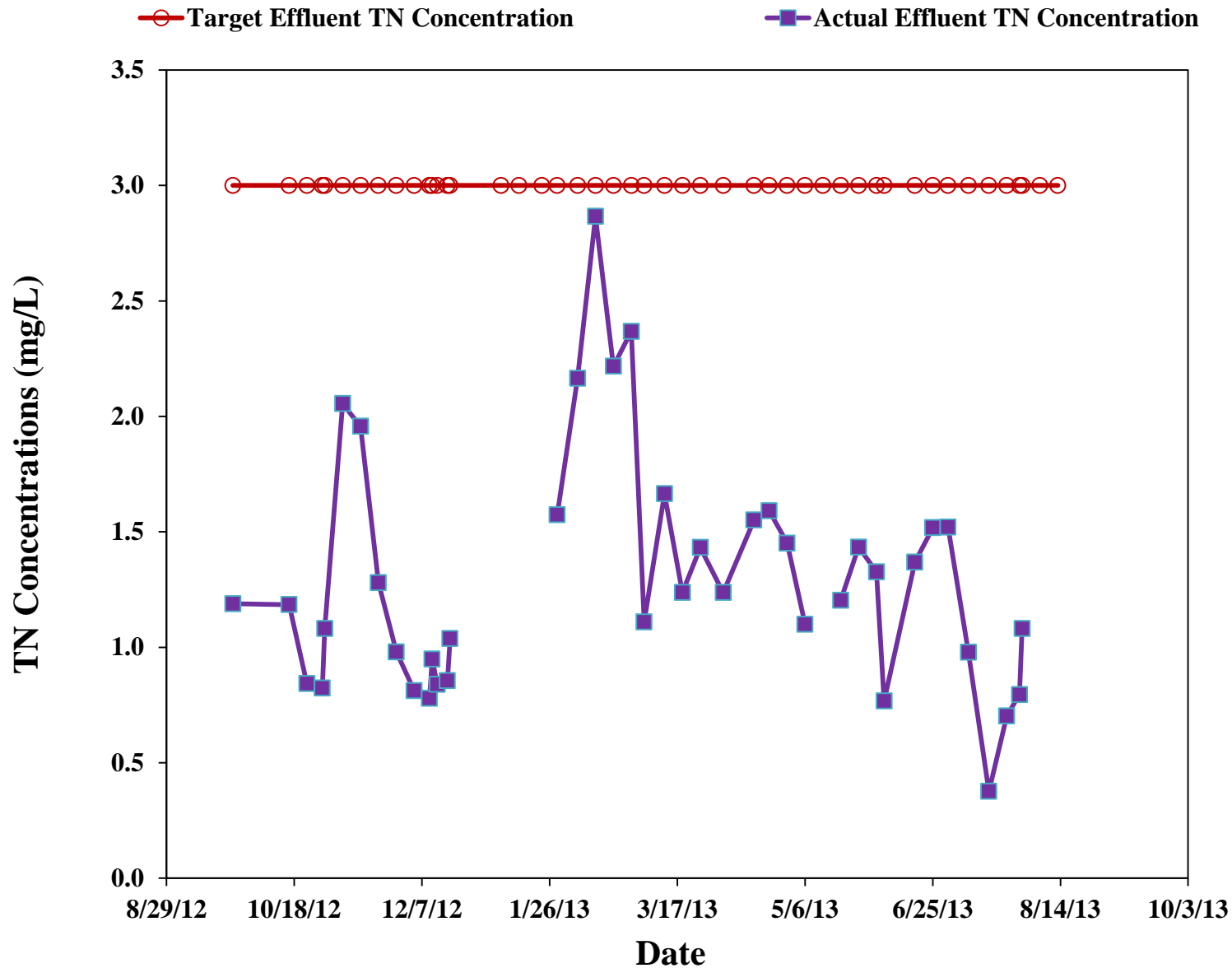
Sources of rbCOD

$$1. \text{COD}_{SO, \text{Decay}} = (1-f_D) \times b_H \times X_H \times \text{HRT}$$

$$2. \text{COD}_{so, \text{Hydrolysis}} = K_H \times \eta_{fe} \times \frac{K_{NO3}}{K_{NO3} + S_{NO3}} \times \frac{X_S/X_H}{K_X + X_S/X_H} \times X_H \times V$$

$$\text{DN Potential} = \frac{\text{COD}_{so, \text{Hydrolysis}}}{2.86/(1-Y_{ANX})} + \frac{\text{COD}_{SO, \text{Decay}}}{2.86}$$





Take-home Message

1. The IFAS processes have established themselves as a well-proven, robust and compact processes for nutrient removal)(now altogether > 600 plants in > 50 countries – 60 installations in the USA).
2. The IFAS is an excellent technology for upgrading existing CAS systems within the existing tankage for either maintaining nitrification at higher flow rates/loads or upgrading a CAS plant to meet new nitrification or total nitrogen removal requirements.
3. The IFAS technology could achieve extremely low $\text{NH}_3\text{-N}$ and $\text{NO}_3\text{-N}$ concentrations in secondary effluent. Total inorganic nitrogen (TIN) concentrations as low as 0.5 mg/L are achievable utilizing IFAS technology.
4. The experience with the operation of Cocoa Beach WRF IFAS plant shows the following: (1) the sludge blanket control strategy in the secondary clarifiers is able to achieve TN and TP less than 1 mg/L and 3 mg/L in effluent in most cases if the sludge blanket is carefully controlled; (2) this process is not as sensitive to TBOD5/TP ratio in plant influent wastewater as the other EBPR processes; (3) this process does not impact the total process SRT selection.



the water quality event

Acknowledgement

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